OILY WASTEWATER TREATMENT BY ULTRAFILTRATION: PILOT-SCALE RESULTS AND FULL-SCALE DESIGN

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ABSTRACT: A West Virginia company that produces an oily waste decided to upgrade their wastewater treatment facilities to prevent possible ground and surface water contamination. The contaminant of concern was oil/grease. West Virginia University's Department of Civil and Environmental Engineering was asked to perform pilot-scale studies using the following treatment technologies: chemical-addition-dissolved air flotation, ultrafiltration, biological filter, constructed wetlands, and land application. Based on data collected over a 2-year period, the following treatment train was proposed: settling ponds \rightarrow oil/water coalescer \rightarrow tubular ultrafiltration \rightarrow constructed wetlands. The ultrafiltration effluent had an oil/grease content that was consistently below 100 mg/ L and very low total suspended solids. Ultrafiltration effluent will be sent to a 15-acre hybrid constructed wetlands/land application system for tertiary treatment. The ultrafiltration residual (2,600 gal/day, >5% oil) will be concentrated by a factor of 10 (260 gal/day, >50% oil) using a high-shear rotary ultrafiltration system. The highly concentrated oil waste will be disposed of via an off-site oil recycler.

INTRODUCTION

In August 1993, a West Virginia company that produces a waste mixture of coolants and lubricants decided to upgrade their industrial wastewater treatment facilities to prevent possible ground and surface water contamination. Until this time, the company collected a mixture of coolants and lubricants from their milling operations and mixed them with more dilute waste in two lined ponds having a combined capacity of about 5 million gallons. A significant amount of the lubricant oils separated and were skimmed from the ponds. The resulting pond effluent, which contained approximately 0.2-0.5% oil/ grease (O/G) and had a daily flow of 80,000 gallons, was then sprayed on a 36 acre site, where the remaining O/G was degraded naturally. In 1993, the West Virginia Department of Environmental Protection issued new ground water regulations that strongly discouraged direct land application of industrial wastes and, shortly thereafter, the company began to investigate alternative treatment methods. West Virginia University's Department of Civil and Environmental Engineering (WVU-CEE) was asked to perform pilot-scale studies for the company involving the following treatment technologies: chemical-addition-dissolved air flotation (CA-DAF), tubular ultrafiltration (UF), biological aerated filter, constructed wetlands, and land application. The primary pollutant of concern from a regulatory standpoint was (O/G).

CA-DAF

CA-DAF is a two-step process. In the first step, a chemical(s) is added to the system to break the oil emulsion and alter the surface chemistry of the particles so that smaller **par**-

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ticles can agglomerate into **larger** particles. In **the flotation** chamber, particles are physically separated from the liquid **us**ing buoyant forces. **Very fine** air bubbles are introduced into the wastewater and are attached to the particle surface. The particle-air bubble system will rise to the surface where **it** can be efficiently removed using **a** skimming apparatus.

Experimental Design

In this study, chemicals from two companies were investigated. Calgon Corp. supplied a cationic polymer (W-2923) to break the emulsion and an anionic polymer (POL-E-Z 2706) to enhance coagulation. KLAR-AID 2400, a cationic polymer, was supplied by Grace Dearborn (GD). Jar tests were used to determine the required chemical dosage prior to CA-DAF Unit operation (Zhu et al. 1997). In Fig. 1, a schematic of the CA-DAF system is presented. The cationic polymers were added directly to the chemical mix tank (CMT). The anionic polymer (Calgon only) was added at the transfer pump inlet and the turbulence from the pump provided the mixing energy. The CA-DAF unit was operated in either the semibatch or continuous mode. In the semibatch mode, the CMT was filled with wastewater and the appropriate amount of cationic polymer (determined by jar tests, see Zhu et al. 1997) and mixed for about 20 min. The contents of the CMT were then transferred to the DAF unit (Calgon anionic polymer added continuously). During the continuous operation the CMT was **constantly be**ing emptied/filled and all chemical additions were continuous. The turbidity of the DAF effluent were measured hourly. If the DAF effluent turbidity was greater than 30 nephelometric turbidity units (NTU), a jar test was conducted to determine if a change in chemical dosage was warranted. Influent and effluent O/G were measured periodically. The flow rate was 8.9 gal/min, recycle ratio = 0.7, and air pressure was 42 psi. The detention times of the CMT and DAF unit were 30 and 67 min, respectively.

Results

DAF effluent turbidity and O/G for a typical **Calgon run** are presented in Fig. 2. **Turbiditys** were highly variable and exceeded the maximum measurable value (200 NTU) on numerous occasions. When the DAF effluent turbidity exceeded 30 NTU a "quick" jar test was performed to determine if the chemical dosage required adjustment. A total of 16 of these tests were conducted during the 2/13-2/19 run. In six of the jar tests a chemical adjustment was required. The new chemical dosage is **indicated in Fig. 2** by the vertically oriented numbers

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FIG. 1. Schematic of CA-OAF System

(e.g., **600/50** \equiv cationic polymer dose/anionic polymer dose). The vertical arrows with no number designation refer to the quick jar tests in which a change in chemical dosage was not warranted. It would typically take 2-3 h before the change in chemical dosage would take effect. Influent O/G ranged from 1,490 to 3,830 mg/L, with an average value of 3,130. Effluent O/G ranged from 11 mg/L to 278 mg/L (60 \pm 80) in the semibatch mode and from 8 to 327 mg/L (205 \pm 135) in the continuous mode. During the semibatch operation, between about 15 and 50 h, the DAF O/G ranged from 12 to 29 mg/L and averaged about 19 mg/L. Despite numerous chemical dosage adjustments, this was the only period of time where consistent and relatively low values of O/G were obtained

DAF effluent turbidity and O/G for a typical GD run are presented in Fig. 3. Turbiditys were lower and less variable compared with results from Calgon. For the DAF effluent, **turbiditys** were generally less than 20, except for a few spikes. The most noticeable turbidity spike occurred at about 70 h. At this time there was a power outage and the chemical **addition** pump failed. The units were idle for about 12 hours and the turbiditys were high following the restartup of the operation, but decreased to less than 30 NTU for the remainder of the run. Turbiditys were slightly higher during continuous operation. Only two jar tests were conducted during the GD run, and for both tests a change in chemical dosage was not warranted. Influent O/G ranged from 2,360 to 3,240 mg/L and averaged 2,950 ± 260 mg/L. Effluent O/G ranged from 15 **mg/L** to 37 **mg/L** (28 \pm 8) in the semibatch mode and from 9 to 41 mg/L (30 \pm 9) in the continuous mode.

Summery

In Table 1 a summary of CA-DAF results are presented. **O**/ G concentrations were significantly better and frequent jar testing was not required for the GD chemical. However, depending on the mode of operation, there was between two and four times more sludge produced for the GD chemical compared with the Calgon chemicals. Assuming a sludge disposal cost ² of **\$0.26/gal**. (supplied by company), the disposal cost per 1,000 gallons of wastewater treated ranged from \$1.77 to **\$1.90 for** the Calgon chemicals and \$3.46 to \$5.98 for the GD **chemical**.

UF

UF is a pressure driven membrane technique that uses porous membranes for the separation of material in the 1 nm-10 µm size range or compounds with molecular weights in excess of 5,000. Membranes that are used for UF are characterized by the molecular weight of a compound that is not able to pass through the membrane [molecular weight cutoff (MWCO)]. In theory, a significant number (90%) of compounds having a molecular weight greater than the MWCO would be retained by the membrane. Membranes having a surface charge equal to the charge of the contaminant can be used The repulsive force between the membrane surface and contaminant decreases membrane fouling. In this study, the pilotscale system was operated first in the semibatch and then in the batch mode. In the semibatch mode, the permeate and raw waste flow rates were equalized by maintaining a constant volume in the feed tank. In the batch mode, no raw waste is added to the system and the concentrate remaining from semibatch operation is "batchdowned." Flow based concentration factors (CF) are calculated as follows:

Semibatch operation:

$$CF_{SB} = 1 + V_{perm}/V_{feed task}$$

Batchdown operation: $CF_{PD} = CF_{PD}$

$$CF_{BD} = CF_{SB} \times [V_{feed \ tank} / (V_{feed \ tank} - V_{perms})]$$

where $CF_{SB} = CF$ during semibatch operation; $V_{perm} =$ volume of permeate produced (gal); $V_{feed tank} =$ volume of the feed tank (gal); and $CF_{BD} = CF$ during batch-down operation. CFs are expressed as 1 X, 2X, etc., and increase with treatment time.

Experimental Design

UF experiments were conducted during the summer (eight runs) and winter (four runs) months. Two membrane types, denoted "**M**" and "P", were investigated (Reed et al. **1997a**). The M membrane had a MWCO of 100,006 and no net surface charge, whereas the P membrane had a MWCO of 120,000 and a net negative surface charge. Both membranes were **con**-structed of **polyvinylidane** fluoride. Sixteen membrane tubes,

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eight for each membrane type, were housed on the UP unit. The average transmembrane pressure was held constant at 42 psi. Samples of **influent** and UF concentrate and **permeate** were **taken** periodically and analyzed for turbidity, O/G, total suspended solids **(TSS)**, and **pH**. Permeate flux was monitored using a graduated cylinder and stopwatch. At the conclusion of batch operation, membranes were cleaned for a total of **60** min with a proprietary cleaning solution.

Results

In Fig. 4, the permeate flux and concentration factor versus time during semibatch operation for summer run 1 (S1) and for winter run 3 (W3) are presented. The horizontal line represents the average permeate flux over the duration of the run The permeate flux versus time for the other UF runs were similar in shape except for summer run 4, which had to be ended early because of low fluxes for both membranes. For all runs, the permeate flux decreased dramatically during the first several hours of operation and then leveled off for the remainder of the semibatch operation. The decrease in flux with time can be attributed to concentration polarization. In

Table 2, a summary of UF data is presented. Run S4 was omitted because the flux decreased to **almost** zero after only 2 h of operation. In a full-scale operation, waste that is processed at such a low flux would be returned to the facility's separation ponds for a longer opportunity to separate and homogenize. The overall average P membrane flux was significantly higher than the M membrane flux [38 versus 27 gal/ft² per day (gpd)], as well as the flux observed during both summer and winter operations. On an individual run basis, the P membrane flux was greater than or equal to the M membrane flux for all runs except W3 and W4. The higher flux for the P membrane can be attributed to its higher MWCO (120,000 compared with 100,000) and negative surface charge. The larger the MWCO the higher the membrane's permeability and the repulsion of the negatively charged O/G droplets by the P membrane surface decreased the fouling layer thickness. The average temperature of the UF influent and concentrate was about 10 degrees lower in the winter than in the summer. A portion of the difference between summer and winter fluxes is due to viscosity differences. Influent O/G ranged from 920 to 5,600 mg/L and averaged 2,460, whereas the TSS ranged from





TABLE 1. Summary of DAF Unit Operation Results for Calgon and GD Chemicals

	Effluent O/G		Sludge Produced		Chemic	al Costs	Sludge Disposal	
	(mg/L)		(gal./1,000)		(\$US/1 ,	000 gal.)	(\$0.26/ga l.)	
Chemicals	Semibatch	Continuous	Semibatch	Continuous	Semibatch	Continuous	Semibatch	Continuous
(1)	(2)	(3)	(4)	(5)	(6)	(7)	(8)	(9)
Calgon GD	60 ± 80 28 ± 8	$205 \pm 135 \\ 30 \pm 9$	7.3 13.3	6.8 23.0	4.54 10.80	4.05 10.80	$\begin{array}{c} 1.90\\ 3.46\end{array}$	3.46 5.98

150 to 2,100 **mg/L** and averaged 645. Waste variability over the course of entire project as well as within a given run was large and demonstrates the complex nature of the settling pond **waste**.

Although there were some **effluent** samples with high **O/G** and TSS concentrations, the vast majority of the samples had O/G concentrations of less than 50 mg/L and TSS levels of less than 25 mg/L. O/G removal efficiencies averaged 98% for the M membrane and 97% for the P membrane. TSS rejections were approximately 97% for both membranes. For all runs, **the** effluent O/G concentration and turbidity **from** the P

membrane were higher than **the** M membrane. Effluent TSS was greater for the P membrane for **all** runs except W3, and S7 TSS for run S7 were almost equal. Given the larger MWCO **of the** P membrane (120,000 compared with 100,000) **the** effluent should be better for the M membrane. Effluent **turbiditys** were measured for all samples in which O/G analyses were performed to determine if there was a relationship between the two parameters. It was hoped that turbidity could be used as a real-time indicator of effluent quality, specifically O/G concentration. However, there was not a **statistically** significant relationship between **O/G** and turbidity.

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FIG. 4. Permeate Flux and CF for Typical Summer (a) and Winter (b) Run8

Batchdown was not conducted for runs **S4**, **S5**, S7, **W1** and W2. For runs S7 and W2, the feed tank overflowed because of a faulty solenoid valve. High temperatures prevented **batch**-down during run S5, whereas low fluxes at the end of the semibatch portion of runs S4 and **W1** necessitated membrane cleaning. If batchdown was not possible in the full-scale operation, the concentrate at the end of semibatch operation would be sent back into the separation ponds to allow for free oil separation and for a more homogenous mixture of wastes to be introduced to the UF unit. CF between 16 and 116X and volume reductions between approximately 94 and 99% were observed at the end of batchdown. Residual production ranged from 10 to 58 gal./1,000 gal. of wastewater treated. Batchdown for run S3 ended prematurely because of high temperatures producing the lowest CF and the highest residual production

(16X and 58 gal/1,000 gal.). The average volume reduction and residual production for the entire project were 97% and 32 gal./1,000 gal., respectively. During batchdown for runs S7, s8, W1, W3, and W4, free oil appeared in the feed tank. The free oil was periodically removed and later analyzed for water content. Free oil production ranged from 2.5 to 7.3 gal./1,000 gal. of wastewater treated. All samples had a trace amount of water associated with the free oil (<5%). The facility in question conducted bumability tests and reported that the free oil was suitable for burning in on-site boiiers.

A preliminary design was conducted for the P membrane using winter conditions. The P membrane was chosen because during winter and summer testing the P membrane flux was consistently higher than that of the **M** membrane, whereas the effluent quality for the P membrane was not significantly **dif**-

TABLE 2. Summary of UF Results

	Average Flux (gai./ft ² per day)		Average O/G (mg/L)		Average TSS (mg/L)]	Volume reduction	Residuals	
Run number (1)	M (2)	P (3)	M (4)	P (5)	M (6)	Р (7)	CF (8)	(%) (9)	(gal./1,000) (10)	
S1 S2 S3 S5 S6 S7 S8 W1 w 2 w 3	26 30 58 16 17 26 32 25 22 22 26	43 63 58 25 20 31 69 29 47 15	39 9 245 25 23 14 26 40 a7	45 11 15 470 55 30 15 34 33 115	NA NA 10 39 NA 6.3 NA 15 7 5 93	NA NA 11 43 NA 6.0 NA 19 5.5 43	$ \begin{array}{c} 27X \\ 23X \\ 16X \\ \hline 60 \\ \hline 116X \\ \hline 42x \\ 42x \end{array} $	95.6 95.6 94.2 96.6 98.3. 97.8	44 44 58 30 18 	
w 4 Average summer Average winter Overall average	17 29 22 27	16 44 26 38	55 40 52 46	82 67 66 66	14 18 32 20	18 20 21 20	104x 	99 	10 	



FIG. 5. Schematic of Land Application Cell (Adapted from Cooper and Hobson 1989)

ferent **than** that observed for the M membrane. The design flow rate is 80,000 gpd. The required membrane area for winter conditions, assuming a 5% expansion capability, is 3,240 ft². During summer months, the system will be able to **process more waste than the** daily flow, which is attractive for the facility because the settling ponds can be drawn down creating additional storage space for periods of inclement weather.

BIOLOGICALAERATEDFILTER

The pilot-scale biological aerated filter consisted of four cylindrical reactors operated in series. Each reactor was 6 ft high and 2.5 ft in diameter. The particle size of the media ranged **from 1/4** inch to 1 inch, and was tapered because of the high solids loading. Air was applied through medium coarse diffusers at **the** bottom of each reactor. A nutrient solution, consisting of ammonium sulfate, nonbasic sodium phosphate, and

dibasic potassium phosphate, was injected into the first reactor. The influent flow rate was varied from 0.5 to 3 gal./min. The hydraulic loading ranged from 294 to 880 gpd/ft², whereas the recirculation ratio ranged **from** 0 to 3. The system was back; washed based on head loss during the first 6 months of operation, whereas a time schedule was developed for the latter part of the study. The wastewater parameters chosen to evaluate the system's performance were O/G and TSS. These parameters were sampled on a daily basis because of the high variability of the waste stream. Effluent-dissolved oxygen, pH, and water temperature were measured during each sampling event. Other analytical testing included chemical oxygen demand, potassium, phosphate, and the various forms of nitrogen. Many difficulties were encountered during the operation of the system mostly due to high solids and O/G loading, insufficient water velocity during backwash events, and low dissolved oxygen. Modifications were made to the system increasing system performance. Removal efficiencies decreased

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with increasing hydraulic loading, whereas recirculation adversely affected system performance. Frequent clogging of the **filter** was problematic and the inability to effectively backwash the filter led to its abandonment as a treatment option.

LAND APPLICATION

A schematic of land treatment cell is presented in Fig. 5. To evaluate the effect of vegetation, two land treatment cells were constructed. One cell was vegetated with a mixture of tall fescue and orchard grass, chosen for their ability to evapotranspirate and hardiness. The seed was allowed to germinate and develop root and leaf structure prior to any wastewater application. This cell was referred to as the V cell. The other cell remained nonvegetated and was referred to as the NV cell. Each cell was 9 ft wide, 29 ft long, and contained soil to a depth of 20 in. Beneath this soil was a geotextile filter fabric and 3 in. of river gravel. The entire system was bounded on the bottom and sides by a **30-mil** polyvinyl chloride (**PVC**) liner. The **effluent** from the system was collected in the river gravel bed and transmitted to two 250-gal. conical bottom tanks. The cells were open to the atmosphere and subject to atmospheric changes. During spraying, a 0.5-A buffer zone around the perimeter of the cells was maintained to minimize short circuiting by the PVC liner. Each week, 0.5 in. of wastewater was evenly applied over the sprayed area. The wastewater application was followed by a **6-day** resting period to restore aerobic conditions. This cycle continued for 1 full year. Influent sampling occurred on the day of application. Effluent sampling occurred 1 day after a spraying and/or significant rain event, and continued on a daily or every **2-day** cycle. in Fig. 6, influent and effluent O/G concentrations are presented for both cells. O/G removal was greater than 97% over the length of the study. There was little difference in performance between the vegetated and nonvegetated cells other than an increase in water loss due to evapotranspiration in the vegetated cells. In a full-scale system, vegetation would be included because of the increase in evapotranspiration and operational considerations (minimizing erosion, wind-blown soil, runoff, etc.).

CONSTRUCTED WETLANDS

A schematic of a constructed wetland cell is presented in Fig. 7. In this study, two wetlands series were constructed with two cells in series (cell $1 \rightarrow$ cell 2; cell $3 \rightarrow$ cell 4). Each cell was 14 ft wide and 26 ft long and had an average depth of 20 in. Cells 1 and 4 had cattails planted aud cells 2 and 3 had bulrush. The ability of the constructed wetlands to polish the effluent from either the CA-DAF or UF system (secondary treatment systems) was evaluated using pond effluent that was diluted with tap water. O/G, total organic carbon, chemical oxygen demand, biochemical oxygen demand, TSS, and dissolved oxygen were measured along the length of each cell; however, only O/G results will be presented here. In Table 3, O/G concentration for constructed wetland series 1 (ceils 1 and 2) are presented. Results for series 2 (cells 3 and 4) were similar. Initially, O/G removal was not effective, but as the vegetation matured O/G removal increased. There was also a noticeable increase in O/G during the colder months. Overall O/G removal efficiency was about 94% with the lowest efficiencies (~88%) occurring in the winter months.

RESIDUAL TREATMENT

Oil skimmings **from** primary treatment (settling ponds) were disposed of through an oil **recycler**. Residuals **from** tertiary treatment (constructed wetlands/land application) consist only of vegetative harvests. Thus, the primary residual requiring



FIG. 6. Influent (a) and Effluent (b) O/G Concentration from Land Application Cells

treatment was from the **secondary** treatment **system** (either CA-DAF or UF). Acid cracking of **CA-DAF and UF residuals** was not effective because the amount of acid required was similar to the amount of effluent produced. A high-shear rotary UF membrane system was tested to determine **the** efficacy of concentrating secondary treatment residuals. The high-shear rotary UF system is described in greater detail in Reed et al. (1997b). In Fig. 8, the permeate flux versus concentration factor is presented for a 5% oil waste. Five percent oil represents a typical oil content from secondary treatment. The high-shear rotary **UF** system was very effective in concentrating the oily waste high oil contents. Concentrate **from** the rotary **UF** system will be disposed of off-site while the permeate from the system.

PRELIMINARY DESIGN

The existing settling ponds will be used for storage and to remove free oil. Effluent from the ponds will then pass through an oil-water separator prior to entering the UF system. A **UF**



FIG. 7. Schematic of Constructed Wetland

TABLE 3. O/G Concentrations for Constructed Wetland Series 1: Cells 1 and 2

	Celi 1 Influe				Cell 1 Efftuen	t	Cell 2 Effluent			
Month (1)	Average (2)	Minimum (3)	Maximum (4)	Average (5)	Minimum (6)	Maximum (7)	Average (8)	Minimum (9)	Maximum (10)	
May June July August September October November December January	209 231 298 162 60 98 233 158 131	47 26 63 90 25 17 69 44 43	319 633 672 407 109 223 342 264 216	56 34 16 12 8 9 23 20 10	36 11 2 3 1 3 5 11 <1	89 68 44 30 26 15 53 36 15	59 20 8 5 3 6 19 20 6	42 S 4 <1 1 9 10 <1	88 39 16 14 8 17 45 35	
February	185	- 4 5 11	618	9	1	23	8	4	18	



FIG. 8. Permeate Flux versus Concentration Factor for 5% Oil Using High-Shear Rotary UF System

tubular membrane system having membranes with a **MWCO** = 120,000 will be used; 3,100 ft^2 of membrane (1,410 tubes) is required. The UF will run until the concentration factor of 32 is reached. At this point, the **UF** system will be shutdown and cleaned and the UF residual (-2,600 gpd) sent to the high-shear rotary UF system (50 ft^2 of membrane) for further

concentration. The rotary system will concentrate the residual by about a factor of 10, resulting in approximately 260 gpd of concentrated oily waste that will be disposed of via an offsite oil **recycler**. Effluent from the UF system will be applied to a constructed **wetlands/land** application hybrid system using land mass from the existing land treatment system. The **exist**ing land treatment application devices will be used to apply **the UF effiuent** in such a manner as to maintain near saturated conditions and an aerobic environment. Bulrush, tall **fescue**, and orchard grass will be planted throughout the system. Required land **areas** for winter and summer conditions are 15 and 5 acres, respectively.

APPENDIX. REFERENCES

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